## **U.G. CHEMISTRY PROGRAMME SPECIFIC OOTCOMES:**

(a) Developing knowledge of chemistry through theory and practical.
b) Explaining molecules in terms of nomenclature, stereochemistry, structures, reactivity,
and mechanism of the chemical reactions.
(c) Identify chemical formulae, structures and solve numerical problems.
(d) Understanding good laboratory practices and safety. (e) Developing research-oriented
skills.
(f) Make aware and handle the sophisticated equipment and advanced instruments.

## **U.G CHEMISTRY COURSE OUTCOME:**

Title		Course Outcome
Chemis	try UG (CBCS) Semeste	r
Organic Chemistry	pictures of bonding. CO-2: To know structure and and factor that control their resonance, hyperconjugation CO-3: To understand MO the CO-4: To learn the basics corron chemical bonding and Phy CO-5: To know the concept examples of elimination, from the pricyclic reactions. CO-6: To understand about the intermediates and their electric CO-7: Gain basic knowled molecules. CO-8: To learn experimentate compounds from solid bir laboratory reagents. CO-9: To understand experimentation of organic liquid Componic of organic liquid Componic control of componic control o	ory and frontier MOs (FMO). Incepts of organic chemistry specially visical Properties. In types, reaction, mechanism and re-radical, nucleophilic substitution are formation and stability of reaction ophilic and nucleophilic behaviour. In ge of Stereochemistry of organically about the separation of organical nary mixture by using common mentally how to determine the boiling ounds.  Ally the qualitative detection of pure
Physical Chemistry-1	Students acquired knowledg Pressure, Temperature, Volun	e on some physical parameter like ne, frequency
	Chemis Organic Chemistry	Chemistry UG (CBCS) Semeste  CO-1: To know hybridisatio pictures of bonding.  CO-2: To know structure an and factor that control their resonance, hyperconjugation CO-3: To understand MO the CO-4: To learn the basics cor on chemical bonding and Phy CO-5: To know the concepexamples of elimination, from and pericyclic reactions.  CO-6: To understand about the intermediates and their electric CO-7: Gain basic knowled molecules.  CO-8: To learn experimentate compounds from solid bin laboratory reagents.  CO-9: To understand experimentate of organic liquid Compounds and liquid organic compounds are constant and liquid organic compounds and liquid organic compounds are constant and liquid organic constant and

Paper/	П	<b>Fitle</b>	Course Outcome
Semester		4 HG (CDCC) C	
	Chemis	stry UG (CBCS) Semester	
CC2 (Theory & Practical Sem-1	Physical Chemistry-1	like Pressure, Temperature, Voc CO2: Learnt about Collision, distribution of molecular spetween ideal and real gas and RMS speed on the basis of M CO3: They also learned to equation from speed distribution CO4: Students are able to expegas and its virial form. Also, like critical temperature, criticoefficient etc.  CO4: Students are learned at their concept by understanding they also learned some new close system, adiabatic system CO5: In thermochemistry part and dissociation of molecule listening about bond energenthalpy effected by changes CO6: Students learned about Entropy change of systems an and transformations; Entropy CO7: Students experience thermodynamic derivation listenment and order of a chemical determine the order (1st, 2nd, 3 CO9: Concept of different king consecutive reaction.  CO10: Students learn Arrhed Lindermann theory of Unimo CO11: Determination of Placolormatching method.  CO12: Determination of placolormatching method.  CO13: Study of kinetics of decental consecutive of the strong base.  CO14: Study of kinetics of decental consecutive of the strong base.  CO14: Study of kinetics of decental consecutive of the strong base.  CO14: Study of kinetics of decental consecutive of the strong base.	binary collision, Collision diameter beed of gas molecules. Difference decay and calculation of MPS, Average speed axwell equation.  The establish the energy distribution ion equation.  The establish the energy distribution is the energy of the energy, and the energy, the energy of

Title Course Outcome		Course Outcome		
Chemistry UG (CBCS) Semester				
Inorganic Chemistry-1	CO-2: To study in detail about chemical properties of the factors influence those proper effect.  CO-3: To understand acid-bath CO-4: To know the basic control of the cont	at modern periodic table, physical and elements along a group or period, rties, relativistic effects and inert pair use reactions.  The property of Red-Ox reactions.  The of ions or salts by acid-base titration		
Organic Chemistry-2	presence of Stereo-axis; cor conformations of stereoisome CO-2: To understand reaction and tautomerism of Organic of CO-3: To learn the concep- equilibria and HSAB principle CO-4: To learn in detail Stereochemistry of substitution CO-5: To learn experimenta	n kinetics, reaction thermodynamics compounds. t of organic acids, bases, acid-base le. about the types, mechanism and		
Physical Chemistry-2	viscosity of different fluid, CO-2: Learn the application usage of electrode kinetics, their reactions, conductance conductance measurement.  CO-3. Learning basic Thermodynamic functions for Learning about chemical thermodynamic condition. CO-4: Learning the spectre molecule, moment of inertia energy of a molecule, zero- Raman scattering, fluorescen mechanism, Operators used	ng technique of surface tension and an of electrochemistry, develops the electrochemical cell formation and and its importance. Application of knowledge of thermodynamics. It systems with variable composition. Equilibrium and derivation under coscopy and structural property of the mode of vibration and vibrational point energy a harmonic oscillator, ance and phosphorescence and their in quantum mechanics, matter and wave function and energy of 1D, 2D, stem.		
	Chemis Inorganic Chemistry-1 Organic Chemistry-2	Chemistry UG (CBCS) Semeste  CO-1: To know extra nuclear CO-2: To study in detail about chemical properties of the factors influence those properties. CO-3: To understand acid-bath CO-5: To study the estimation method and Oxidation –Reduted and Experimental Stereochemistry of Substitution (CO-2: To understand reaction and tautomerism of Organic CO-3: To learn the concept equilibria and HSAB principing CO-4: To learn in detail Stereochemistry of substitution (CO-5: To learn experimental yield and determine the melting in the laboratory.  CO-1: Learning the measuring viscosity of different fluid, CO-2: Learn the application usage of electrode kinetics, their reactions, conductance conductance measurement.  CO-3: Learning basic Thermodynamic functions for Learning about chemical thermodynamic condition. CO-4: Learning the spectromolecule, moment of inertial energy of a molecule, zero-Raman scattering, fluorescent mechanism, Operators used wave property of a system, we wave property of a system.		

Paper/	Title		Course Outcome
Semester Chemistry UG (CBCS) Semester			
CC5 (Theory & Practical Sem-III	Physical Chemistry-2 Practical	Learn about viscosity meaning respect to water. Partition distribution of I2 between where a strong, weak/monobasic, of Study of saponification reasonable.	surement of unknown liquid with coefficient determination for the vater and CCl <sub>4</sub> and learn about titration of an acid dibasic) against base strong action conductometrically n of Ostwald's dilution law and
CC6 (Theory & Practical Sem-III	Inorganic Chemistry-2	bonding, laws, rules and edbonds, solubility, hybridization CO-2: To study the moder (Molecular Orbital Theory, weak intermolecular forces) CO-3: To understand about radioactive compounds, nucl radio carbon dating, hazards CO-4: To study the estimation antimony in a mixture, availated Iodo-Iodimetry Titration Met	on of copper in Brass, chromium and
CC7(Theory & Pract6ical)	CC-7(Organic Chemistry-3)	reactions and reaction mecha CO-2: To understand about nucleophilic aromatic s intermediates and their mech CO-3: To study the prop compounds and correspondin CO-4: To learn preparations, mechanisms of organometalls	different types of electrophilic and substitution reactions, reaction anisms.  Derties and reactions of carbonyling reaction mechanisms.  Treactions and corresponding reaction ic compounds.  The importance of the compounds of the compounds of the compounds.
SEC-1 (Theory & Practical)	SEC-1(Pharmaceutical Chemistry)	development of representati Antipyretic, Analgesics,	about the drug discovery, design and live drugs of the following classes: Anti-inflammatory, Anti-bacterial, otics, Anti-leprosy, Central Nervous elated drugs.

Paper/ Semester	Title	Course Out Come
	Chemistry UG (C	CBCS) Semester
SEC-1 (Theory& practical. SEM-III	SEC-1(Pharmaceutical Chemistry)	CO-2: To know about aerobic and anaerobic fermentation, importance of Vitamins and Amino acids, synthesis of Penicillin, Cephalosporin, Chloromycetin, Streptomycin and their role as an antibiotic.  CO-3: To learn experimentally how to prepare aspirin in the laboratory and how to analyze it.  CO-4: To learn experimentally how to prepare magnesium bisilicate in the laboratory.
Sem-IV CC-8(Theory& Practical)	CC-8(Physical Chemistry-3)	Learning about application of Thermodynamics in types of solution, colligative property and determination of molecular mass of solutes using colligative properties.  Phase rule and its derivations, learn about three component systems.  Aquired knowledge on binary solutions, Electrical Properties of molecules. Application of standard electric potential in different kind of half-cell  Schrödinger equation, transformation. Know about quantization of energy. Hartree-Fock method development.
	Practical	Learning about potentiometric titration of Mohr's salt solution. Know about phenol-water phase diagram. pH-metric titration.
Sem-IV CC-9(Theory& Practical)	CC-9 (Inorganic Chemistry-III)	CO-1: To study the general principles of metallurgy. CO-2: To study the chemistry of s and p block elements including noble gases and their compounds in detail.  CO-3: To learn about inorganic polymers in detail.  CO-4: To study the meaning of various terms involving co-ordination chemistry, Warner" s theory for complex formation, structural and stereoisomerism of co-ordination complex.  CO-5: To study the estimation of Zn (II), Zn (II) in a Cu (II) and Zn (II) mixture, Ca(II) and Mg (II) in a mixture, Hardness of water by Complexometric Titration Method.  CO-6: To follow the preparation methods of [Cu (CH <sub>3</sub> CN) <sub>4</sub> ] PF <sub>6</sub> /ClO <sub>4</sub> , Cis and trans K [Cr (C <sub>2</sub> O <sub>4</sub> ) <sub>2</sub> (H <sub>2</sub> O) <sub>2</sub> ], Potassium di aqua dioxalato chromate (III), Tetrammine carbonato Cobalt (III) ion, Potassium tris (Oxalato) ferrate (III), Tris-(ethylenediamine)  Nickel (II) chloride, [Mn(acac) <sub>3</sub> ] and [Fe (acac) <sub>3</sub> ].

Paper/ Semester	Title	Course Out Come
Chemistry UG (CBCS) Semester		
SEM-IV CC-10 (Theory& practical.	CC-10(Organic Chemistry-IV	CO-1: To understand in detail about the synthesis, separation, properties, identification, chemical reactions and their corresponding mechanism of nitrogen containing compounds.  CO-2: Discussion about different kinds of rearrangement reactions.  CO-3: Helps to know the logic of organic synthesis
		CO-4: To study UV-Visible, IR and NMR spectroscopy in detail.
SEM-IV Sec-2 Theory& Practical)	(Pesticide Chemistry)	CO-1: Helps to understand about the preparation, structures, properties, reactions, benefits and adverse effects of pesticide compounds CO-2: Helps to understand how to calculate acidity/alkalinity in a given sample of pesticide formulations as per BIS specifications. CO-3: To learn experimentally how to prepare organophosphates, phosphonates and thiophosphates. CO-4: To study how to prepare inorganic complexes in the laboratory. CO-5: To know how to determine the co-ordination compounds by spectrophotometric method.
	Cosmetic Chemistry	CO-1: Formulate and evaluate various cosmeceutical products. CO-2: Know the key components used in different cosmeceutical products. CO-3: Recognize the role of ingredients and herbs used in cosmeceutical products. CO-4: Know the advanced current technology used for manufacturing the cosmetics at lab scale and industry scale. CO-5: Employable skills and high technical competence for Pharceutical industry
	Fuel Chemistry	Learning about renewable and non-renewable energy sources. Petroleum and non-petroleum fuels. Petrochemicals, Lubricants.

Paper/ Semester	Title	Course Out Come
Chemistry UG (CBCS) Semester		
SEM-V CC-11 (Theory& practical	CC-11(Inorganic Chemistry-IV	CO-1: To study the concept of VBT, elementary idea of CFT, OSSE, MO concept, Sigma-Pi bonding in complex, Magnetism and Colour og complexes, Orgel diagram, Selection rules for Spectral studies.  CO-2: To study the chemistry of d and f- block elements and their compounds in detail.  CO-3: To study the principles and techniques involved in chromatographic separation of Ni (II) and Co (II), Fe (III) and Al (III) by Paper chromatography.  CO-4: To learn the Gravimetric estimation of Ni (II), Al(III), Cu as CuSCN, chloride.  CO-5: To study the measurement of 10 Dq and λ max of [Mn (acac) <sub>3</sub> ] and [Fe(acac) <sub>3</sub> ] by Spectrophotometry Method.
SEM-V CC12 (Theory & Practical)	CC1-12(Organic Chemistry-V)	CO-1: To learn in detail about the synthesis, properties, chemical reactions and reaction mechanisms of polynuclear hydrocarbons and their derivatives.  CO-2: To study the chemical reactions, properties and synthesis of heterocyclic compounds.  CO-3: To know in detail about the stereochemistry, properties and chemical reactions of alicyclic compounds.  CO-4: To learn the mechanism, stereochemistry and regioselectivity of pericyclic reactions.  CO-5: Helps to understand about the classification, structure, properties, reactions and use of carbohydrate molecules.  CO-6: Deals with the synthesis, structure, properties, chemical and biological reactions of amino acids, peptides and nucleic acids.  CO-7: To learn experimentally how to separate molecules by chromatographic methods  CO-8: To study how to analyze the Organic compounds by spectroscopic techniques.

SEM-V DSE-1 (Theory& practical.	DSE-1 (T) Advanced Physical Chemistry	CO-: -Learning about solid states, type of unit cell, packing efficiency, plane of symmetry of crystal, Bragg's reflection.  To understand the differences between statistical thermodynamics and previously learned thermodynamics.  Applications to barometric distribution. Maxwell's speed distribution.  3rd law of thermodynamics and their consequence, Classification of polymers, Mechanism and kinetics of polymerization.
	Practical	Computer programs based on numerical methods for Programming. Like roots of equations, Numerical differentiation.  Application of Gauss-Siedel method in colourimetry.  Uses of molecular visualization software.
DSE-2 (Theory & Practical)	DSE-2T Analytical Methods in Chemistry	CO: Learning about Sampling, evaluation of analytical data, errors, accuracy and precision, statistical test of data; F, Q and t test, rejection of data, and confidence intervals.  CO: Learned about Origin of spectra, Basic principles of instrumentation, Basic principles of quantitative analysis, Infrared Spectrometry, Classification of electroanalytical methods. Separation techniques like Solvent extraction, Mechanism of extraction, Technique of extraction. chromatography, Role of computers in instrumental methods of analysis.
	Practical	To learn the separation technique by Chromatography (TLC), Analysis of soil: (i) Determination of pH of soil. (ii) Estimation of calcium, magnesium, phosphate.

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SEM- VI CC-13 (Theory & Practical)	CC-13(T) Inorganic Chemistry-V	co-1: To study the benefits and role of essential elements and metal ions in biological system, structure bonding and mechanism of Haemoglobin, Hemocyanin, Hemerythrin, electron transport protein Hydrolytic enzymes, mechanism of Photosystem (I, II), effects of toxic metals, role of Pt and Au as drugs and metal dependent diseases.  co-2: To study the definition and classification of Organometallic compounds, preparation, reactions and structures of carbonyls, nitrosyls, cyanides, Zeise's salt.  co-3: To know about the catalytic role of Organometallic compounds in different types of industrial processes.  co-4: To study the reaction kinetics and mechanism of different type of complexes
	Practical	.CO-5: To study experimentally the qualitative detection of known and unknown radicals and insoluble materials in a mixture.
SEM- VI C14T: (Theory & Practical)	CC14 (T) Physical Chemistry-V	CO: 1. To learn about fundamentals and application of Rotational, Vibrational, Raman & NMR Spectroscopy. Principle and applications of ESR. CO:2 Learning Photochemistry calculation of Quantum yields, Photophysical and Photo chemical Process.
		CO:3. Learned about Surface Chemistry Adsorption, Colloids etc.
	Practical	lem:lem:lem:lem:lem:lem:lem:lem:lem:lem:
DSE-3 (Theory & Practical	Green Chemistry (T)	CO:1 Learning about Green Chemistry Principles, Synthesis, how green Chemistry gives us a sustainable environment  CO:2 Future Trends in Green Chemistry
		CO:2 Future Trends in Green Chemistry

DSE-3 (Theory & Practical	Practical	Preparation and characterization of nanoparticles of gold using tea leaves  Preparation of biodiesel from vegetable/ waste cooking oil
DSE4T: (Theory & Practical)	Polymer Chemistry	CO: To learn classification of polymers.  Mechanism and kinetics of copolymerization.
		Morphology of crystalline polymers